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Modelling the long-term effect of irrigation with gypsiferous water on soil and water resources

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Abstract

The use of gypsiferous mine water for irrigation of agricultural crops is a promising technology that could solve problems related to both shortage of irrigation water and disposal of effluent mine drainage. The long-term effect of irrigation with lime-treated acid mine drainage on soil properties and catchment salt load was investigated. The soil water-salt balance-crop growth model (SWB) and the CLIMGEN weather data generator were used to simulate 30 years of irrigation with gypsiferous mine water for different irrigation management scenarios, followed by 20 years of dry land summer cropping, to determine if the problem of salt disposal was merely being postponed. Generated weather input data were for Bethal (South Africa), soil input data for a Plinthic Ferralsol (FAO-UNESCO) and crop input data for a rotation of pearl millet (Pennisetum glaucum cv. SA Standard) and oats (Avena sativa L. cv. Overberg). The soil appeared to act as an effective salt sink, with large quantities of calcium sulphate (340-404 Mg ha⁻¹) being precipitated in 30 years, and with negligible amounts of remobilization thereafter. Dissolved salts in the soil solution increased during winter when rainfall is negligible. The highest concentrations occurred deeper in the profile, which is desirable because rooting is less dense there and plant growth is thus less affected. Due to large amounts of gypsum precipitation near the surface, the ions in solution were mainly Mg^{2+} and SO_4^{2-} . Simulated root weighed, saturation electrical conductivities indicated that summer cropping should present no problems, but for certain winter crops a leaching fraction should be applied to obtain maximum yields. Drainage water quality was variable, depending heavily on rainfall. Peak salt levels were around 9.6 g L⁻¹. Depending on irrigation strategy, between 418 and 636 Mg ha⁻¹ of salts was leached over 30 years. Once irrigation ceased, very little leaching occurred. Annual irrigation depended on rainfall, but averaged at about 1100 mm per annum. About 350 mm of percolation occurred, resulting in net utilization of around 750 mm annually. About 40% of the salts added through irrigation could be immobilized in the soil profile. It is concluded that year round, high frequency irrigation, with a leaching fraction in winter, should be an effective and economical means of utilizing large quantities of gypsiferous water without causing irreparable damage to soil resources. ©1999 Elsevier Science B.V. All rights reserved.

Keywords: Acid mine drainage; Gypsiferous water; Irrigation; Model studies; Soil salinity; Weather data generator

1. Introduction

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The mining industry is one of the most important industries in South Africa, both from the points of

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view of gross national product and job creation. Mining is spread over a wide region, including important high-potential agricultural areas. In the mining of mineral resources, pollution problems are created, with adverse effects on the already scarce water resources (66% of the country is classified as semi-arid to arid with low and variable rainfall). Add to this the steady increase in population, and the maintenance and effective utilization of agricultural potential and water resources becomes of paramount importance (Department of Water Affairs and Forestry, 1997).

Most South African coal and gold deposits contain pyritic formations. When exposed to oxygen, water and the catalytic action of *Thiobacillus ferrooxidans* bacteria, iron pyrite is oxidized to sulfuric acid and iron sulphate (Thompson, 1980). This results in large quantities of acid mine drainage (AMD) being formed. The extremely high acidity of this water, with pH typically ranging from 2 to 4.4 (Jovanovic et al., 1998), prohibits discharge into natural streams, as the environmental impact would be severe. This is not a local problem, but occurs worldwide where similar deposits are found, including old underground workings (Pulles et al., 1995).

Current measures to prevent pollution of the environment were summarized by Pulles et al. (1996). In several mines, AMD is treated with hydrated lime, Ca(OH)₂, in order to neutralize acidity. Most of the resulting CaSO₄ is precipitated in sedimentation basins, but the effluent water is saline (gypsiferous), with an electrical conductivity (EC) typically in the range from 130 to 290 mS m⁻¹, due mainly to the presence of Ca²⁺ and SO₄²⁻ in solution (Jovanovic et al., 1998). If freely discharged into the natural environment, lime-treated AMD and similar waters emanating from mines could cause salinization of soils, rivers, dams and catchment areas.

The utilization of mine waste water must be balanced with the conservation of soil and water resources in such a way that the resources and associated ecosystems are not degraded to such an extent as to render them useless and unproductive. Lime-treated AMD has previously been used for dust alleviation on dirt roads and for irrigation of lawns. The possible utilization of this water for irrigation of agricultural crops shows some promise (Du Plessis, 1983a). Large amounts of waste water could possibly be utilized for irrigation of high potential soils. Moreover, filtering

saline water through the soil and thereby precipitating gypsum in the profile, could limit environmental pollution. The high cost of AMD amelioration could also be offset to some extent by farming income.

The feasibility of using mine water for irrigation of agricultural crops was evaluated by Barnard et al. (1998), in a three-year field screening trial carried out at Landau Colliery Kromdraai Opencast Section (near Witbank, Mpumalanga Province, South Africa). For the purpose of lime-treated AMD utilization, they recommended fast growing species that use a lot of water. They concluded that irrigation with lime-treated AMD should not present a soil salinity or crop production problem within a relatively short time period. The problem of saline water management, however, has both short- and long-term implications (Bresler et al., 1982). Long-term changes in soil chemical properties can be measured, but the execution of long-term experiments monitoring slow environmental processes can be prohibitively expensive and it is not always possible to wait for the outcome of such experiments. Computer models, combined with a weather data generator, are useful tools for predicting environmental consequences of management practices.

Several crop production–salt movement models have already been published (Maas and Hoffman, 1977; Bresler et al., 1982; Du Plessis, 1983b; Wagenet, 1984; Campbell, 1985). Most of them, however, are either crop specific or do not simulate the mechanism of daily crop water uptake. In this study, a simple dynamic soil water–salt balance-generic crop growth model, SWB (Annandale et al., 1996a), was used to assess the long-term impact of irrigation with gypsiferous mine water on soil and water resources for a location in South Africa. Long-term weather records were generated using the CLIMGEN weather data generator of G.S. Campbell (Washington State University), a modified version of WGEN (Richardson and Wright, 1984).

The SWB model and CLIMGEN were used to evaluate the following factors for different scenarios of irrigation water management: (1) Quantities of salts (CaSO₄) that are likely to be immobilized within the irrigated soil profile over time; (2) The probable influence of the deposition of salts on soil porosity; (3) Salt concentrations in the soil profile, over time; (4) The expected quality of the water draining from the irrigated area during the period of irrigation, and following the

cessation of irrigation using mine water; (5) Indication of the influence of drainage water on catchment salt load; (6) The impact of variations in seasonal and annual rainfall on the soil and drainage water characteristics; and 7) Water and salt balance.

2. Materials and methods

2.1. Description of the model

Soil water balance (SWB) is a mechanistic, multi-layer, daily time step, soil water–salt balance-generic crop growth model, developed from NEWSWB, a modified version of the model published by Campbell and Diaz (1988). A flow diagram for SWB is presented in Fig. 1.

A cascading soil water balance is used once canopy interception and surface runoff have been accounted for. Each soil layer is assumed to fill to field capacity and then pass on any remaining water to the layer below. Any water that passes beyond the bottom layer is assumed lost to deep percolation. Potential evapotranspiration (PET) is calculated as a function of daily average air temperature, vapor pressure deficit, radiation and wind speed, adopting the internationally standardized FAO Penman-Monteith methodology (Smith et al., 1996). The two components of PET potential evaporation and potential transpiration — are estimated from canopy cover. Actual transpiration is determined on a daily basis as the lesser of root water uptake or maximum loss rate. Total soil water potential is used to determine the amount of water available for crop transpiration in each soil layer. The osmotic effect on crop growth is simulated by adding osmotic potential to the matric and gravitational soil water potentials. Osmotic potential is calculated as a function of ionic concentration (Campbell, 1985). The daily dry matter increment (DM_i) is taken as the minimum of the water supply limited (Tanner and Sinclair, 1983) and radiation limited DM_i (Monteith, 1977). A stress index, the ratio between actual and potential transpiration, is used as a limiting factor for canopy development.

Required input data are planting date, latitude, altitude, rainfall and irrigation water amounts and quality, as well as maximum and minimum daily temperature.

In the absence of measured data, SWB estimates solar radiation, vapor pressure and wind speed according to the FAO recommendations (Smith, 1992). In addition, volumetric field capacity and permanent wilting point, initial volumetric soil water content and the content of ionic species (Ca²⁺, Mg²⁺, Na⁺, K⁺, Cl⁻ and SO₄²⁻) are required for each soil layer. Since SWB is a generic crop growth model, the following crop specific parameters have to be experimentally determined in order to accurately simulate the soil water balance: daily radiant transmission coefficient, vapor pressure deficit, corrected dry matter water ratio, radiation conversion efficiency, day degrees required for crop developmental stages, cardinal temperatures, dry matter partitioning parameters, specific leaf area and stress index. The SWB model includes a database of growth parameters for a wide range of crop and pasture species.

Chemical precipitation/dissolution of lime and gypsum is calculated on a daily time step per soil layer, using the model published by Robbins (1991). Critical assumptions of the chemical equilibrium model and the validation with lysimeter data are discussed in the same study. The model of Robbins (1991) solves chemical equilibrium by iteration. Within each iteration, activity coefficients and ion activities are calculated for Ca²⁺, Mg²⁺, Na⁺, H⁺, SO₄²⁻, HCO₃⁻ and CO₃²⁻, and the solution phase is equilibrated with solid phase lime and gypsum. EC is calculated from individual ion concentrations (McNeal et al., 1970) for each soil layer. The SWB model ends the iteration procedure when the change in EC between the previous and the following loop is <0.01 mS m⁻¹.

The model of Robbins (1991) requires an input value for CO_2 partial pressure (pCO_2) in soil air. The values of pCO_2 can vary widely in cultivated soils, depending on soil physical properties, root depth and aeration. Its effect on soil chemistry is, however, very limited in the salinity range interesting for irrigation with mine water (Oster, 1990). A constant pCO_2 value of 0.3% (or kPa) is used in SWB. Atmospheric pCO_2 is 0.03% (Reuss, 1991).

The SWB model calculates only concentrations of soluble ionic species and does not include the cation exchange procedure of Robbins (1991). It was assumed that Ca²⁺ would quickly saturate the soil exchange complex while irrigating with gypsiferous water. Soil solution–solid phase interactions were

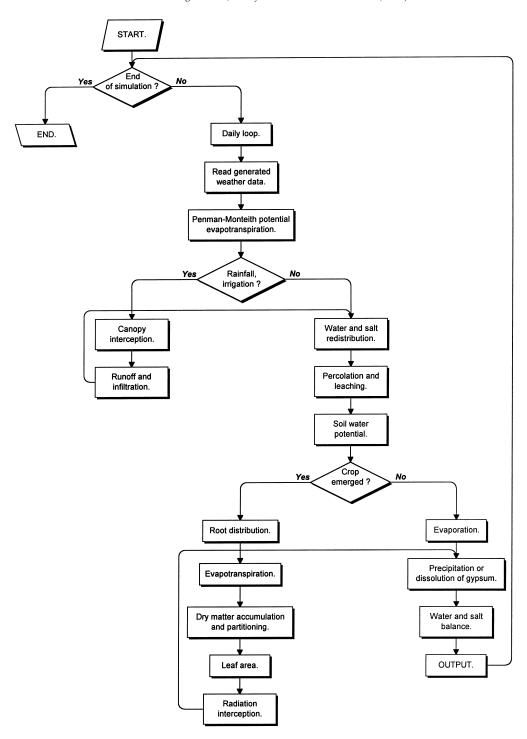


Fig. 1. Flow diagram of the soil water balance (SWB) model.

therefore not expected to significantly affect the parameters that determine crop water use and soil salinity in the long term. The introduction of the cation exchange subroutine would likely make the model more reliable for short-term simulations, but more inputs would then be required.

Weather data are generated using CLIMGEN, a two-component weather generator. The first component (CLIMPAR) determines statistical parameters required for long-term daily weather data generation, while the second component (CLIMGEN) generates weather data. CLIMGEN generates daily maximum ($T_{\rm max}$) and minimum air temperature ($T_{\rm min}$), as well as precipitation (P) from either historic daily weather data, if available, or from monthly averages. At least 20 years of rainfall and 10 years of historic temperature data are required in order to get a reliable simulation. The CLIMGEN weather data generator was assessed at South African sites by Clemence (1997), who showed the estimations to be quite satisfactory.

The SWB model is written in Delphi and runs in Windows 95. The Fortran code published by Robbins (1991) was translated and included in a user-friendly format in SWB. A copy of the model's code and the model are available from the authors.

2.2. Input parameters for long-term predictions

Long-term simulations were carried out for a rotation of pearl millet (Pennisetum glaucum cv. SA Standard) and oats (Avena sativa L. cv. Overberg) grown for forage, thereby providing high soil water consumption rates throughout the year. Crop parameters for pearl millet and oats were calibrated using data from the field trial carried out at Landau Colliery Kromdraai Opencast Section (Barnard et al., 1998). In that trial, optimal agronomic techniques for the area were implemented. The crops were sprinkle irrigated with lime-treated AMD and showed no signs of foliar injury. For the long-term simulations, SWB was modified to simulate the harvesting of oats. This was done by reducing the calculated value of total dry matter production (TDM) to 0.5 Mg ha⁻¹, on days when TDM was equal to, or greater than, $3.5 \,\mathrm{Mg}\,\mathrm{ha}^{-1}$. SWB simulated the re-growth of the crop thereafter, until the planting of pearl millet.

Long-term simulations were carried out for an unmined site, located in the area of the Kleinkopje opencast coal mine (near Witbank). The soil is loamy sand (810 g kg⁻¹ sand, 50 g kg⁻¹ silt and 140 g kg⁻¹ clay), with a plinthic layer below 1.5 m. It is classified as Bainsvlei form–Redhill series according to the Soil Classification Working Group (1991), or as Plinthic Ferralsol (FAO-UNESCO, 1974). Soil physical and chemical properties reported by De Villiers (1987) were used as input in the model. Soil water content at field capacity and permanent wilting point were calculated as a function of soil texture, according to the equations recommended by Bennie et al. (1988).

The following irrigation water quality data were obtained from Amcoal (Landau Colliery) and used as model input: $\text{Ca}^{2+} = 0.013 \, \text{mol} \, \text{L}^{-1}$; $\text{SO}_4{}^{2-} = 0.021 \, \text{mol} \, \text{L}^{-1}$; and $\text{Mg}^{2+} = 0.008 \, \text{mol} \, \text{L}^{-1}$. This water emanates from old underground workings and needs to be disposed of in order to allow planned future opencast mining. The presence of other ions in the proposed irrigation water, including heavy metals, was neglected as their concentrations were very low. Rain water quality data reported by Bolt (1982) were used as model input.

Twenty years of historic weather data were collected for the meteorological station in Bethal, the nearest to the Kleinkopje mine with a reasonably long record of data, and used as input in the weather data generator. It was assumed that similar weather conditions to those in Bethal would occur in the area surrounding the Kleinkopje mine. Fifty years of daily $T_{\rm max}$, $T_{\rm min}$ and P were generated by CLIMGEN in order to carry out the long-term crop growth and soil salinity simulations. Examples of generated weather data are presented in Fig. 2 (monthly average $T_{\rm max}$ and $T_{\rm min}$) and Fig. 3 (monthly P). Realistic weather data was generated by CLIMGEN compared to the typical weather conditions for the given site.

The SWB model was modified to include options where frequency and amounts of simulated water applications can be chosen. In this study, irrigations were simulated on days when soil water deficit was calculated to be equal to, or greater than, 20 mm. The figure of 20 mm corresponds approximately to one-sixth of the available water in the top 1 m soil layer. The following irrigation strategies were adopted: (1) irrigation amounts refilling the soil profile to field capacity (FC); (2) irrigation amounts replacing two-thirds

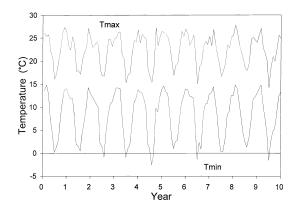


Fig. 2. Monthly average maximum and minimum air temperatures ($T_{\rm max}$ and $T_{\rm min}$) simulated with the CLIMGEN weather data generator for Bethal (South Africa).

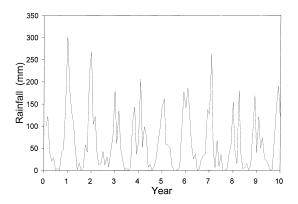


Fig. 3. Monthly rainfall simulated with the CLIMGEN weather data generator for Bethal (South Africa).

of the soil water deficit (0.67 FC); and (3) irrigation amounts applying a 20% leaching fraction (20% LF).

Irrigations were simulated for 30 years, after which another 20 years were simulated for dry land pearl millet grown during the summer rainy season. The purpose was to evaluate the drainage water quality after the cessation of irrigation and to see whether substantial remobilization of precipitated gypsum would occur.

3. Results and discussion

3.1. Quantities of gypsum immobilized in the soil profile over time

The SWB model precipitates or dissolves CaSO₄, depending on solution concentration. This is affected

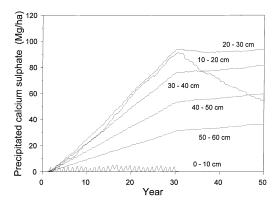


Fig. 4. Cumulative amounts of CaSO₄ (Mg ha⁻¹) precipitated at different depths in the profile of loamy sand Plinthic Ferralsol for a pearl millet-oats rotation irrigated at field capacity for 30 years, followed by dry land cropping.

by the amount and quality of irrigation water, irrigation management and crop water use. In order to get salt mass balance, precipitation/dissolution of calcium sulphate (molecular mass $\approx 136 \,\mathrm{g}\,\mathrm{mol}^{-1}$) was calculated, instead of precipitation/dissolution of the twice hydrated gypsum molecule (molecular mass $\approx 172 \,\mathrm{g} \,\mathrm{mol}^{-1}$). Fig. 4 represents the predicted amounts of CaSO₄ (in Mg ha⁻¹) precipitated at different depths in the soil profile of a pearl millet-oats cropping system for the FC strategy. Except for the surface layer, CaSO₄ was predicted to precipitate mostly in the top soil layers, where the soil solution becomes increasingly concentrated due to water uptake by plants. A slow process of CaSO₄ dissolution by rainfall was predicted after the cessation of irrigation. This results because dissolution in the upper layers is accompanied by an increased precipitation in the lower layers. This means that large quantities of salt could be permanently removed from the water system. The soil appears, therefore, to act as an effective salt sink. Depending on irrigation strategy, between 11.35 and 13.47 Mg ha⁻¹ per year of CaSO₄ was predicted to be immobilized in the soil profile after 30 years of irrigation with this gypsiferous water (Table 1). As can be expected, more precipitation occurred for the deficit irrigation strategy (0.67 FC), as the soil solution tended to be more concentrated and the irrigation frequency higher.

Table 1
Simulated values of water and salt balance parameters during 30 years of irrigation with mine water for three irrigation management strategies (Kleinkopje mine, Witbank, South Africa)

Parameter	Irrigation strategy		
	0.67 FC ^a	FCb	20% LF ^c
Evapotranspiration (mm per year)	1486	1394	1394
Total irrigation (mm per year)	1047	1019	1222
Total added salts (Mg ha ⁻¹ per year)	28.62	27.84	33.40
Total drainage (mm per year)	261	325	529
Total leached salts (Mg ha ⁻¹ per year)	13.93	15.11	21.20
Total precipitated CaSO ₄ (Mg ha ⁻¹ per year)	13.47	11.85	11.35
Maximum decrease in soil porosity (%)	11.8	10.2	9.7
TDS^d at 10–20 cm soil depth (gL ⁻¹)	2.7-9.8	2.7–7.1	2.7-6.0
TDS^d at 100–110 cm soil depth (g L ⁻¹)	3.2-9.6	3.0-7.5	2.8-5.8
Maximum $TDS^{d)}$ in drained water (g L^{-1})	9.6	7.5	5.7

^a Irrigation amounts replacing 2/3 of the soil water deficit.

3.2. Influence of salt deposition on soil porosity

The present form of SWB is unable to predict fertility changes as N, P, K and micronutrients are not simulated. This is not seen as a serious disadvantage for practical applications, as it is relatively easy to monitor in the field and correct any nutrient deficiencies that may arise over time due to Ca²⁺ dominating the exchange complex. Soil physical effects, however, are of greater concern, as these may not always be easy to remedy. Low concentrations of Na⁺ in AMD and the saline conditions should ensure a good structure, so infiltration is expected to be unimpeded by surface crusting. The only possible concern is that of CaSO₄ precipitation clogging soil pores. This could reduce aeration and hydraulic conductivity. Pore clogging can be simulated by calculating the volume that the precipitated CaSO₄ will occupy. The volume of CaSO₄ (V) clogging soil pores in each layer was calculated as follows:

$$V = \frac{\text{Mass CaSO}_4}{\text{Density CaSO}_4} \tag{1}$$

The density of CaSO₄ was assumed to be $2.3\,\mathrm{Mg\,m^{-3}}$. The density of soil solid particles was assumed to be $2.65\,\mathrm{Mg\,m^{-3}}$ and volumetric soil porosity (P_s) was calculated as follows:

$$P_{\rm s} = 1 - \left(\frac{\rho_{\rm b}}{2.65}\right) \tag{2}$$

where ρ_b is the bulk density of the soil (Mg m⁻³). The decrease in soil porosity (P_{sd}) per unit area was then calculated:

$$P_{\rm sd} = \frac{V}{P_{\rm s} \, dz} \tag{3}$$

where dz represents the depth of soil layer in meters. Model predictions indicated that a reduction of pore volume up to 11.8% could occur in the top soil layers after 30 years of irrigation with gypsiferous water (Table 1). Precipitation of CaSO₄ was predicted to occur mostly in the surface layers where cultivation should be an effective remedy to pore clogging. More accurate simulations would result from a model based on Richard's equation/Darcy's law as one could then bring in the effect of pore clogging on hydraulic con-

3.3. Salt concentrations in the soil profile over time

ductivity.

The SWB model has a multi-layer soil component; so, solution salt concentrations of the various layers are predicted over time. Total dissolved salts (TDS) in the soil solution are expected to increase during the dry winters, while a decrease in soil salinity is to be expected during the rainy summer seasons. Predicted TDS values were mainly dependent on the occurrence of rainfall, and did not show any particular trend during the 30-year simulation. After the cessation of irrigation, TDS approached $2.64 \, \mathrm{g \, L^{-1}}$, the experimentally

^b Irrigation amounts refilling the soil profile to field capacity.

^c Irrigation amounts applying a 20% leaching fraction.

^d Total dissolved salts.

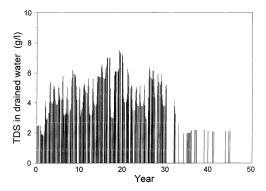


Fig. 5. Simulated total dissolved salts (TDS) in drainage water for Plinthic Ferralsol and pearl millet—oats rotation irrigated at field capacity for 30 years, followed by dry land cropping.

determined value of a saturated CaSO₄ solution (Bolt and Bruggenwert, 1978). Higher TDS values were predicted to occur at depths below the root system where salts are flushed by irrigation water and rainfall. Lower and more stable TDS values were generally calculated when applying a leaching fraction (Table 1).

3.4. Expected percolate quality during and after irrigation with mine water

The model keeps track of the amount and quality of water percolating below the rooting depth (1.1 m). Drainage water quality was highly dependent on rainfall and salt dynamics in the soil profile. The irrigation strategy followed had a noticeable effect, with the deficit irrigation resulting in a more concentrated percolate (Table 1). Total dissolved salts in drained water peaked at around 9.6 g L⁻¹ for the deficit or 0.67 FC strategy, equivalent to a solution electrical conductivity of about 1300 mS m⁻¹ (US Salinity Laboratory Staff, 1954). It was calculated that TDS in percolating water would have a decreasing trend after the cessation of irrigation and in most years no leaching would occur (Fig. 5).

3.5. Effect of drainage water on catchment salt load

The total mass of salts leached can be determined from the quality and quantity of the leachate integrated over the irrigated area. In Table 1, salts leached during the 30-year simulation are presented for the three irrigation strategies. An average of between 13.93 and

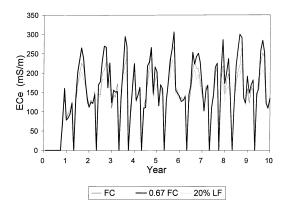


Fig. 6. Simulated root weighted soil solution electrical conductivity at saturation (EC_e) for Plinthic Ferralsol, pearl millet—oats rotation and three irrigation management strategies (FC: field capacity, LF: leaching fraction).

21.20 Mg salts per hectare per year, mostly MgSO₄, can be expected to be leached over 30 years, depending on water management. The deficit irrigation strategy resulted in the least leaching, and the leaching fraction strategy, the most.

3.6. Impact of annual and seasonal variations in rainfall on soil and drainage water quality

The levels of TDS in the soil solution are expected to increase during the dry season due to irrigation with saline water, whilst lower TDS values were calculated in summer. More concentrated soil solutions during dry periods cause less favorable conditions for crop growth. Values of electrical conductivity at saturation (EC_e) were calculated per soil layer, as a function of EC and soil water content. Profile root weighted EC_e was calculated from each soil layer ECe and the fraction of the total root length in that layer (Campbell and Diaz, 1988). Root weighted EC_e values are shown in Fig. 6. Data refer to a 10-year simulation for a pearl millet-oats cropping system and three irrigation management strategies. Root weighted ECe is equal to 0 when no roots are present in the soil profile. According to the guidelines of Maas (1986), summer conditions are favorable for growth of all crops regardless of irrigation strategy. High values of root weighted EC_e, especially for the deficit irrigation strategy, were predicted in winter, when it seems prudent to apply a leaching fraction in order to achieve optimal yields.

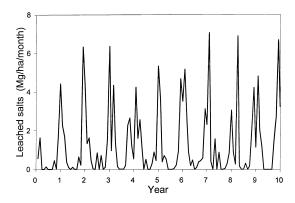


Fig. 7. Predicted salt leaching (Mg ha⁻¹ per month) for Plinthic Ferralsol and pearl millet-oats rotation irrigated at field capacity.

As a result of the higher soil solution concentrations during the dry period of the year, CaSO₄ is expected to precipitate mostly in winter. During the rainy season, conditions for limited dissolution of CaSO₄ will materialize. The critical period of the year from the point of view of salt leaching is summer (Fig. 7).

3.7. Water and salt balance

Simulations with SWB gave valuable estimates of the volume of water that could be used and mass of salts that could be disposed of, by adopting different management practices. This can therefore assist the planning of the required area to be irrigated, should this be deemed to be a feasible option for utilizing such water. Considerable volumes of gypsiferous water could be used by the cropping system simulated (Table 1). It is interesting to note that the deficit irrigation strategy requires approximately the same amount of irrigation water as the FC strategy. This is occurring because of the greater direct evaporation component of the deficit irrigation strategy, caused by a higher frequency of irrigation. Average crop evapotranspiration is 1400 mm per annum. Annual irrigation depends on rainfall, but averages out at about 1100 mm per annum, or 11 000 m³ ha⁻¹. In practice, this may be slightly less if logistical problems are encountered with removing an old crop and planting a new one timeously. On average, about 350 mm of percolation occurs resulting in a net utilization of around 750 mm annually.

Table 1 reports the predicted total amounts of salts added through irrigation, precipitated in the form of

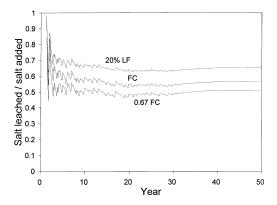


Fig. 8. Predicted salt leached/salt added ratio for Plinthic Ferralsol, pearl millet—oats rotation and three irrigation management strategies (FC: field capacity, LF: leaching fraction).

CaSO₄ and leached out of the soil profile. About 40% of the salts added through irrigation water could be immobilized in the soil profile. The greatest amount of salts added and leached was predicted for the strategy applying a leaching fraction (20% LF), whilst the largest amount of precipitated CaSO₄ was for the deficit strategy (0.67 FC). Salts intercepted by crop canopies and the salt mass dissolved in the soil solution were calculated to be negligible components of the salt balance over the long term. The 0.67 FC deficit irrigation strategy resulted in the lowest cumulative salt leached/salt added ratio (Fig. 8). A high frequency irrigation strategy may, therefore, be recommended for the purpose of using lime-treated AMD for irrigation of agricultural crops. Similar results have been obtained for simulations with a mine rehabilitated soil profile and lucerne-fescue perennial pasture (Annandale et al., 1996b).

4. Summary and conclusions

The water balance–soil salinity model presented in this work, was specifically developed to solve problems concerning the utilization of lime-treated acid mine drainage. It proved to be a useful tool for predicting the outcome of various irrigation management scenarios.

The model was used to simulate 30 years of irrigation with gypsiferous water, followed by 20 years of dry land summer cropping, to determine if the

problem of salt disposal was merely being postponed. According to the long-term simulations for a pearl millet-oats rotation on a sandy soil, substantial volumes of water can be used, and masses of salt disposed of, through irrigation. This can be best achieved through high frequency irrigation of crops year round. It is important to have as large a transpiring canopy as possible at all times. The soil appears to act as an effective salt sink, with large quantities of calcium sulphate (340–404 Mg ha⁻¹) being precipitated over 30 years and with negligible amounts of remobilization thereafter. Possible pore clogging with precipitated gypsum is not seen as a serious problem as it is expected to occur in the top soil layers, where the soil solution tends to concentrate due to crop water uptake. The accompanying cultivation should open up the soil structure thereby alleviating the problem. Dissolved salts in the soil solution are expected to increase during winter when rainfall is negligible. The highest concentrations occurred deeper in the profile, which is desirable, because rooting is less dense there and therefore plant growth is less affected. Due to large amounts of gypsum precipitation near the surface, the ions in solution were mainly Mg²⁺ and SO₄²⁻. Simulated root weighted ECe indicated that summer cropping should present no problems, but for certain winter crops a leaching fraction should be applied to attain maximum yields. Drainage water quality was variable, depending heavily on rainfall. Peak salt levels were around 9.6 g L⁻¹, equivalent to a solution electrical conductivity of about $1300 \,\mathrm{mS} \,\mathrm{m}^{-1}$. Between 418 and 636 Mg ha⁻¹ of salts, mainly MgSO₄, were predicted to be leached over 30 years. Once irrigation ceased, very little leaching occurred. As far as seasonal variations are concerned, most gypsum precipitated in winter and most leaching occurred in summer. It is concluded that year round, high frequency irrigation, with a leaching fraction in winter, should be an effective and economical means of disposing of large quantities of salt and water without causing irreparable damage to soil resources.

The SWB model should give a good first approximation of the movement of water in well-drained, fairly uniform profiles. One should, however, be aware of the limitations of such soil water balance models. Textural discontinuities and impermeable layers have marked effects on water movement and this can only be reasonably described by a finite-difference

model, which moves water according to Darcy's law (i.e., using water potential and hydraulic conductivity). Another disadvantage of the cascading model is that upward flow of salts and water is not simulated. This could be important especially if a high, or perched, water table is present.

The model could be further developed by introducing a subroutine quantifying aluminum hydroxide interactions in the soil solution. In addition, the simulation of root water uptake under saline conditions could be improved by accounting for decreased root permeability (O'Leary, 1970; Majeed et al., 1994; Ferrer and Stockle, 1996), or local osmotic potentials at the root–soil interface (Stirzaker and Passioura, 1996).

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